SECTION 1: Identification

1.1. Product identifier
Product form : Substance
Name : Nitrogen, refrigerated liquid
CAS No : 7727-37-9
Formula : N2
Other means of identification : Liquid Nitrogen refrigerant R728

1.2. Recommended use and restrictions on use
Recommended uses and restrictions : Medical applications
Industrial use
Manufacture of food products
Food applications

1.3. Supplier
Josef Gas
201 Basaltic Rd
Concord - Canada L4k 1G4
T 416-658-1212
www.josefgases.com

1.4. Emergency telephone number
Emergency number : 1-613-996-6666 CANUTEC.
For routine information, contact your supplier or Josef Gas sales representative.

SECTION 2: Hazard identification

2.1. Classification of the substance or mixture
GHS-CA classification
Simple asphyxiant H380
Refrigerated liquefied gas H281

2.2. GHS Label elements, including precautionary statements

GHS-CA labelling

Hazard pictograms : GHS04

Signal word : WARNING

Hazard statements : CONTAINS REFRIGERATED GAS; MAY CAUSE CRYOGENIC BURNS OR INJURY MAY DISPLACE OXYGEN AND CAUSE RAPID SUFFOCATION

Precautionary statements : Do not handle until all safety precautions have been read and understood
Use and store only outdoors or in a well-ventilated area
Wear cold insulating gloves and either face shield or eye protection
Use a back flow preventive device in the piping
Close valve after each use and when empty
Always keep container in upright position
DO NOT change or force fit connections
<table>
<thead>
<tr>
<th>2.3.</th>
<th>Other hazards</th>
</tr>
</thead>
<tbody>
<tr>
<td>Other hazards not contributing to the classification</td>
<td>: Asphyxiating in high concentrations. Contact with liquid may cause cold burns/frostbite.</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>2.4.</th>
<th>Unknown acute toxicity (GHS-CA)</th>
</tr>
</thead>
<tbody>
<tr>
<td>: No data available</td>
<td></td>
</tr>
</tbody>
</table>
SECTION 3: Composition/information on ingredients

3.1. Substances

Name: Nitrogen, refrigerated liquid
CAS No: 7727-37-9
EC no: 231-783-9

<table>
<thead>
<tr>
<th>Name</th>
<th>CAS No.</th>
<th>% (Vol.)</th>
<th>Common Name (synonyms)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Nitrogen</td>
<td>(CAS No) 7727-37-9</td>
<td>&lt;= 100</td>
<td>Nitrogen (liquified) / NITROGEN</td>
</tr>
</tbody>
</table>

3.2. Mixtures

Not applicable

SECTION 4: First-aid measures

4.1. Description of first aid measures

First-aid measures after inhalation: Remove victim to uncontaminated area wearing self contained breathing apparatus. Keep victim warm and rested. Call a doctor. Apply artificial respiration if breathing stopped.

First-aid measures after skin contact: The liquid may cause frostbite. For exposure to liquid, immediately warm frostbite area with warm water not to exceed 105°F (41°C). Water temperature should be tolerable to normal skin. Maintain skin warming for at least 15 minutes or until normal coloring and sensation have returned to the affected area. In case of massive exposure, remove clothing while showering with warm water. Seek medical evaluation and treatment as soon as possible.

First-aid measures after eye contact: Immediately flush eyes thoroughly with water for at least 15 minutes. Hold the eyelids open and away from the eyeballs to ensure that all surfaces are flushed thoroughly. Contact an ophthalmologist immediately. Get immediate medical attention.

First-aid measures after ingestion: Ingestion is not considered a potential route of exposure.

4.2. Most important symptoms and effects (acute and delayed)

No additional information available

4.3. Immediate medical attention and special treatment, if necessary

Other medical advice or treatment: None.

SECTION 5: Fire-fighting measures

5.1. Suitable extinguishing media

Suitable extinguishing media: Use extinguishing media appropriate for surrounding fire.

5.2. Unsuitable extinguishing media

No additional information available

5.3. Specific hazards arising from the hazardous product

Reactivity: No reactivity hazard other than the effects described in sub-sections below.
Reactivity in case of fire: No reactivity hazard other than the effects described in sub-sections below.

5.4. Special protective equipment and precautions for fire-fighters

Firefighting instructions: DANGER! Extremely cold liquid and gas under pressure. Take care not to direct spray onto vents on top of container. Do not discharge sprays directly into liquid; cryogenic liquid can freeze water rapidly. Evacuate all personnel from the danger area. Use self-contained breathing apparatus (SCBA) and protective clothing. Immediately cool containers with water from maximum distance. Stop flow of gas if safe to do so, while continuing cooling water spray. Remove ignition sources if safe to do so. Remove containers from area of fire if safe to do so. On-site fire brigades must comply with their provincial and local fire code regulations.

Protection during firefighting: Compressed gas: asphyxiant. Suffocation hazard by lack of oxygen.

Special protective equipment for fire fighters: Use self-contained breathing apparatus. Standard protective clothing and equipment (Self Contained Breathing Apparatus) for fire fighters.

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Nitrogen, refrigerated liquid  
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**Specific methods**

- Use fire control measures appropriate for the surrounding fire. Exposure to fire and heat radiation may cause gas containers to rupture. Cool endangered containers with water spray jet from a protected position. Prevent water used in emergency cases from entering sewers and drainage systems.

- Exposure to fire may cause containers to rupture/explode.

- Stop flow of product if safe to do so.

- Use water spray or fog to knock down fire fumes if possible.

- If leaking do not spray water onto container. Water surrounding area (from protected position) to contain fire.

**Other information**

- Cryogenic liquid causes severe frostbite, a burn-like injury. Heat of fire can build pressure in a closed container and cause it to rupture. Venting vapors may obscure visibility. Air will condense on surfaces such as vaporizers or piping exposed to liquid or cold gas. Nitrogen, which has a lower boiling point than oxygen, evaporates first, leaving an oxygen-enriched condensate.

- Containers are equipped with a pressure relief device. (Exceptions may exist where authorized by TC).

**SECTION 6: Accidental release measures**

**6.1. Personal precautions, protective equipment and emergency procedures**

**General measures**

- Evacuate area. Ensure adequate air ventilation. Wear self-contained breathing apparatus when entering area unless atmosphere is proven to be safe. Prevent from entering sewers, basements and workpits, or any place where its accumulation can be dangerous. Stop leak if safe to do so.

**SECTION 7: Handling and storage**

**7.1. Precautions for safe handling**

- Wear leather safety gloves and safety shoes when handling cylinders. Protect cylinders from physical damage; do not drag, roll, slide or drop. While moving cylinder, always keep in place removable valve cover. Never attempt to lift a cylinder by its cap; the cap is intended solely to protect the valve. When moving cylinders, even for short distances, use a cart (trolley, hand truck, etc.) designed to transport cylinders. Never insert an object (e.g, wrench, screwdriver, pry bar) into cap openings; doing so may damage the valve and cause a leak. Use an adjustable strap wrench to remove over-tight or rusted caps. Slowly open the valve. If the valve is hard to open, discontinue use and contact your supplier. Close the container valve after each use; keep closed even when empty. Never apply flame or localized heat directly to any part of the container. High temperatures may damage the container and could cause the pressure relief device to fail prematurely, venting the container contents. For other precautions in using this product, see section 16.

**7.2. Conditions for safe storage, including any incompatibilities**

- Store in a cool, well-ventilated place. Store and use with adequate ventilation. Store only where temperature will not exceed 125°F (52°C). Firmly secure containers upright to keep them from falling or being knocked over. Install valve protection cap, if provided, firmly in place by hand. Store full and empty containers separately. Use a first-in, first-out inventory system to prevent storing full containers for long periods.

**OTHER PRECAUTIONS FOR HANDLING, STORAGE, AND USE:** When handling product under pressure, use piping and equipment adequately designed to withstand the pressures to be encountered. Never work on a pressurized system. Use a back flow preventive device in the piping. Gases can cause rapid suffocation because of oxygen deficiency; store and use with adequate ventilation. If a leak occurs, close the container valve and blow down the system in a safe and environmentally correct manner in compliance with all international, federal/national, state/provincial, and local laws; then repair the leak. Never place a container where it may become part of an electrical circuit.
SECTION 8: Exposure controls/personal protection

8.1. Control parameters
No additional information available

8.2. Appropriate engineering controls
Appropriate engineering controls: Oxygen detectors should be used when asphyxiating gases may be released. Systems under pressure should be regularly checked for leakages. Provide adequate general and local exhaust ventilation. Consider work permit system e.g. for maintenance activities.

8.3. Individual protection measures/Personal protective equipment

Hand protection: Wear work gloves when handling containers. Wear heavy rubber gloves where contact with product may occur.

Eye protection: Wear goggles and a face shield when transfilling or breaking transfer connections. Select in accordance with the current CSA standard Z94.3, "Industrial Eye and Face Protection", and any provincial regulations, local bylaws or guidelines.

Respiratory protection: Respiratory protection: Use respirable fume respirator or air supplied respirator when working in confined space or where local exhaust or ventilation does not keep exposure below TLV. Select in accordance with provincial regulations, local bylaws or guidelines. Selection should be based on the current CSA standard Z94.4, "Selection, Care, and Use of Respirators." Respirators should also be approved by NIOSH and MSHA. For emergencies or instances with unknown exposure levels, use a self-contained breathing apparatus (SCBA).

Thermal hazard protection: Wear cold insulating gloves. Wear cold insulating gloves when transfilling or breaking transfer connections.

Environmental exposure controls: None necessary.

Other information: Other protection: Safety shoes for general handling at customer sites. Metatarsal shoes and cuffless trousers for cylinder handling at packaging and filling plants. Select in accordance with the current CSA standard Z195, "Protective Foot Wear", and any provincial regulations, local bylaws or guidelines. For working with flammable and oxidizing materials, consider the use of flame resistant anti-static safety clothing.

SECTION 9: Physical and chemical properties

9.1. Information on basic physical and chemical properties
Physical state: Gas
Appearance: Colorless liquid.
Molecular mass: 28 g/mol
Colour: Colourless liquid.
Odour: No odour warning properties.
Odour threshold: No data available
pH: Not applicable.
pH solution: No data available
Relative evaporation rate (butylacetate=1): No data available
Relative evaporation rate (ether=1): Not applicable.
Melting point: -210 °C
Freezing point: No data available
Boiling point: -195.8 °C
Flash point: No data available
Critical temperature: -149.9 °C
Auto-ignition temperature: Not applicable.
Decomposition temperature: No data available
Vapour pressure: Not applicable.
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<table>
<thead>
<tr>
<th>Property</th>
<th>Value</th>
</tr>
</thead>
<tbody>
<tr>
<td>Vapour pressure at 50 °C</td>
<td>No data available</td>
</tr>
<tr>
<td>Critical pressure</td>
<td>3390 kPa</td>
</tr>
<tr>
<td>Relative vapour density at 20 °C</td>
<td>No data available</td>
</tr>
<tr>
<td>Relative density</td>
<td>0.8</td>
</tr>
<tr>
<td>Relative density of saturated gas/air mixture</td>
<td>No data available</td>
</tr>
<tr>
<td>Density</td>
<td>808.5 kg/m³ Liquid density at boiling point and 1 atm</td>
</tr>
<tr>
<td>Relative gas density</td>
<td>0.97</td>
</tr>
<tr>
<td>Solubility</td>
<td>Water: 20 mg/l</td>
</tr>
<tr>
<td>Log Pow</td>
<td>Not applicable</td>
</tr>
<tr>
<td>Log Kow</td>
<td>Not applicable</td>
</tr>
<tr>
<td>Viscosity, kinematic</td>
<td>Not applicable</td>
</tr>
<tr>
<td>Viscosity, dynamic</td>
<td>Not applicable</td>
</tr>
<tr>
<td>Viscosity, kinematic (calculated value) (40 °C)</td>
<td>No data available</td>
</tr>
<tr>
<td>Explosive properties</td>
<td>Not applicable</td>
</tr>
<tr>
<td>Oxidizing properties</td>
<td>None</td>
</tr>
<tr>
<td>Flammability (solid, gas)</td>
<td>Non flammable</td>
</tr>
</tbody>
</table>

**Other information**

- **Gas group**: Refrigerated liquefied gas
- **Additional information**: Gas/vapour heavier than air. May accumulate in confined spaces, particularly at or below ground level

**SECTION 10: Stability and reactivity**

**10.1. Reactivity**

- **Reactivity**: No reactivity hazard other than the effects described in sub-sections below.
- **Chemical stability**: Stable under normal conditions.
- **Possibility of hazardous reactions**: None.
- **Conditions to avoid**: Avoid high temperatures, exposure to Lithium (Li), Neodymium (Nd), Titanium (Ti), Magnesium.
- **Incompatible materials**: None.
- **Hazardous decomposition products**: Under certain conditions, nitrogen can react violently with lithium, neodymium, titanium (above 1472°F/800°C), or magnesium to form nitrides. At high temperature, it can also combine with oxygen and hydrogen.

**SECTION 11: Toxicological information**

**11.1. Information on toxicological effects**

- **Acute toxicity (oral)**: Not classified
- **Acute toxicity (dermal)**: Not classified
- **Acute toxicity (inhalation)**: Not classified
- **Skin corrosion/irritation**: Not classified
  - **pH**: Not applicable.
- **Serious eye damage/irritation**: Not classified
  - **pH**: Not applicable.
- **Respiratory or skin sensitization**: Not classified
- **Germ cell mutagenicity**: Not classified
- **Carcinogenicity**: Not classified
- **Reproductive toxicity**: Not classified
- **Specific target organ toxicity (single exposure)**: Not classified
**Specific target organ toxicity (repeated exposure)**

- Not classified

**Aspiration hazard**

- Not classified

**SECTION 12: Ecological information**

### 12.1. Toxicity

**Ecology - general**
No ecological damage caused by this product.

### 12.2. Persistence and degradability

**Nitrogen, refrigerated liquid (7727-37-9)**
- Persistence and degradability: No ecological damage caused by this product.

**Nitrogen (7727-37-9)**
- Persistence and degradability: No ecological damage caused by this product.

### 12.3. Bioaccumulative potential

**Nitrogen, refrigerated liquid (7727-37-9)**
- Log Pow: Not applicable.
- Log Kow: Not applicable.
- Bioaccumulative potential: No ecological damage caused by this product.

**Nitrogen (7727-37-9)**
- Log Pow: Not applicable for inorganic gases.
- Log Kow: Not applicable.
- Bioaccumulative potential: No ecological damage caused by this product.

### 12.4. Mobility in soil

**Nitrogen, refrigerated liquid (7727-37-9)**
- Mobility in soil: No data available.
- Log Pow: Not applicable.
- Log Kow: Not applicable.
- Ecology - soil: No ecological damage caused by this product.

**Nitrogen (7727-37-9)**
- Mobility in soil: No data available.
- Log Pow: Not applicable for inorganic gases.
- Log Kow: Not applicable.
- Ecology - soil: No ecological damage caused by this product.

### 12.5. Other adverse effects

- Other adverse effects: Can cause frost damage to vegetation.
- Effect on the ozone layer: None
- Effect on global warming: No known effects from this product

**SECTION 13: Disposal considerations**

### 13.1. Disposal methods

**Waste disposal recommendations**
- Dispose of contents/container in accordance with local/regional/national/international regulations. Contact supplier for any special requirements.

**SECTION 14: Transport information**

### 14.1. Basic shipping description

**TDG**
- **TDG Primary Hazard Classes**:
  - UN1977: 2.2-Class 2.2: Non Flammable, Non-Toxic Gas
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Proper shipping name: NITROGEN, REFRIGERATED LIQUID

Explosive Limit and Limited Quantity Index: 0.125 L
Passenger Carrying Ship Index: 450 kg
Passenger Carrying Road Vehicle or Passenger Carrying Railway Vehicle Index: 50 L

14.3. Air and sea transport

IMDG
UN-No. (IMDG): 1977
Proper Shipping Name (IMDG): NITROGEN, REFRIGERATED LIQUID
Class (IMDG): 2.2 - Non-flammable, non-toxic gases
MFAG-No: 120

IATA
UN-No. (IATA): 1977
Proper Shipping Name (IATA): NITROGEN, REFRIGERATED LIQUID
Class (IATA): 2

SECTION 15: Regulatory information

15.1. National regulations
Nitrogen, refrigerated liquid (7727-37-9)
Listed on the Canadian DSL (Domestic Substances List)

Nitrogen (7727-37-9)
Listed on the Canadian DSL (Domestic Substances List)

SECTION 16: Other information

Date of issue: 01/01/2016
Revision date: 04/08/2017

Indication of changes:
Training advice: The hazard of asphyxiation is often overlooked and must be stressed during operator training.

Other information: When you mix two or more chemicals, you can create additional, unexpected hazards. Obtain and evaluate the safety information for each component before you produce the mixture. Consult an industrial hygienist or other trained person when you evaluate the end product. Before using any plastics, confirm their compatibility with this product.

NFPA health hazard: 3 - Short exposure could cause serious temporary or residual injury even though prompt medical attention was given.

NFPA fire hazard: 0 - Materials that will not burn.
NFPA reactivity: 0 - Normally stable, even under fire exposure conditions, and are not reactive with water.
NFPA specific hazard: SA - This denotes gases which are simple asphyxiants.
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HMIS III Rating

Health : 3 Serious Hazard - Major injury likely unless prompt action is taken and medical treatment is given

Flammability : 0 Minimal Hazard - Materials that will not burn

Physical : 2 Moderate Hazard - Materials that are unstable and may undergo violent chemical changes at normal temperature and pressure with low risk for explosion. Materials may react violently with water or form peroxides upon exposure to air.

This information is based on our current knowledge and is intended to describe the product for the purposes of health, safety and environmental requirements only. It should not therefore be construed as guaranteeing any specific property of the product.